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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

## **Assessment Chapter Test B**

Holt McDougal Modern Chemistry 2 Chapter Test Chapter Test A, continued \_\_\_\_ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of their atomic number. c. wave patterns repeat at regular intervals.

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18. Dalton's atomic theory agreed with the modern atomic theory EXCEPT for the statement that all atoms of the same element have the same Complete the following table to compare the types of subatomic particles. 19. 20. 21. 10 Particle Proton Neutron Electron CHAPTER 3 TEST Mass number Relative charge Locatio MODERN CHEMISTRY

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